

# Access Free Electrochemistry Multiple Choice Questions Answers

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electricity through NaCl, the amount of Cl deposited on cathode is. Related: PES Scholastic Aptitude Test Chemistry. A. 17.75 gm. B. 142 gm. C. 71 gm. D. 35.5 gm. Question 2 ...

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this guide covers electrochemistry multiple choice questions (mcqs) with answers and explanation.

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spontaneous reaction that occurs when the



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cell in the picture operates is as follows:  
 $2\text{Ag}^+ + \text{Cd (s)} \rightarrow 2 \text{Ag (s)} + \text{Cd}^{2+}$  (A) Voltage increases. (B) Voltage decreases but remains > zero.

AP Chemistry: Electrochemistry Multiple Choice Answers

SAHOTA 03 Electrochemistry Study Guide - Multiple Choice - Page 1 of 22 1. DO ALL. THE. QUESTIONS in this booklet. These are actual Provincial Exam questions! Your own provincial exam and unit test will include questions similar to the ones in this booklet! 2. RESIST THE. URGE

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THE "OFFICIAL" CHEMISTRY 12 REDOX  
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Multiple Choice Questions (Type-II) Note : In the following questions two or more than two options may be correct. The positive value of the standard electrode potential of  $\text{Cu}^{2+} / \text{Cu}$  indicates that \_\_\_\_\_. (i) this redox couple is a stronger reducing agent than the  $\text{H}^{+} / \text{H}_2$  couple. (ii) this redox couple is a stronger oxidising agent than  $\text{H}^{+} / \text{H}_2$ .

Class 12 Important Questions for Chemistry -  
Electrochemistry

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31. During electrolysis of water the volume of oxygen liberated is  $2.24 \text{ dm}^3$ . The volume of hydrogen liberated, under same conditions will be :  $2.24 \text{ dm}^3$ .  $1.12 \text{ dm}^3$ .  $4.48 \text{ dm}^3$ .  $0.56 \text{ dm}^3$ . Answer. 32. In Down's method for the extraction of sodium, the melting point of the electrolyte is lowered by adding.

Important Questions of Electrochemistry for NEET Chemistry ...

Answers: 1. (d) 2. (e) 3. (a) 4. (e) 5. (d) 6. (a) 7. (d) 8. (b) 9. (d) 10. (c) 11. (b) 12. (c) 13. (b) 14. (c) 15. (c) 16. (d) 17. (e) 18. (b) 19. (e) 20. (c) 21. (c) 22. (d)

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Sample Questions - Chapter 21

If  $E^\circ \text{Fe}^{2+}/\text{Fe} = -0.441 \text{ V}$  and  $E^\circ \text{Fe}^{2+}/\text{Fe}^{3+} = 0.771 \text{ V}$ , the standard EMF of the reaction,  $\text{Fe} + 2\text{Fe}^{3+} \rightarrow 3\text{Fe}^{2+}$  will be. (a) 1.212 V. (b) 0.111 V. (c) 0.330 V. (d) 1.653 V. Answer.

Answer: a. We hope the given Chemistry MCQs for Class 12 with Answers Chapter 3 Electrochemistry will help you.

Chemistry MCQs for Class 12 with Answers Chapter 3 ...

Multiple choice questions Try the following multiple choice questions to test your

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knowledge of this chapter. For each question there is one correct answer. The periodic table, physical constants and relative atomic masses needed for these problems are given on the inside covers of Chemistry, fourth edition by C.E. Housecroft and E.C. Constable.

Multiple choice questions - Pearson Education  
Sample Questions - Chapter 21

Electrochemistry Examples of Multiple Choice Questions: 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the \_\_\_\_\_ ; the chemical

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change that occurs at this electrode is called \_\_\_\_\_. (a) anode, oxidation (b) anode, reduction (c) cathode, oxidation (d) cathode, reduction

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Multiple Choice. Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. The half-reaction that occurs at the cathode during the electrolysis of molten sodium bromide is \_\_\_\_\_. a.  $+ 2e^- 2Br^- \rightarrow Br_2$  b.  $+ 2e^- Br_2 \rightarrow 2Br^-$  c.  $+ e^- Na^+ \rightarrow Na$  d.

AP Chemistry-Electrochemistry

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Mcqs Of Chapter Electrochemistry

Answer. Answer: (b) 2. An increase in the conductivity equivalent of a solid electrolyte with dilution is primarily due to. (a) increased ionic mobility of ions. (b) 100 percent electrolyte ionisation with natural dilution. (c) increase in both ion numbers and ionic mobility. (d) A rise in ion counts.



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MCQ on Electrochemistry - NCERT Books

NEET Chemistry Electrochemistry Multiple Choice Questions MCQs on Electrochemistry :

1. The aqueous solution of which of the following compounds is the best conductor of electric current?

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