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Question Bank Topic 3 Metals Mastering Chemistry Suggested answers to in-text activities and unit-end exercises Topic 3 Unit 10 Practice P10.1(page 8) a) i) Shiny silvery globules of lead ii) lead(II) oxide + carbon lead + carbon monoxide (or carbon dioxide) b) Carbon monoxide c) No reaction occurs Carbon cannot remove oxygen from aluminium oxide. P10.2(page 12) a) Haematite b) i) The coke ...

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Question Bank Topic 3 Metals Mastering Chemistry Heating the oxide of Z gives metal Z. Z is probably a metal low in the reactivity series, such as mercury or silver. 8 A 9 A The more reactive a metal is, the more difficult it is to extract it from its ore. X is a reactive metal as it is difficult to extract from its oxide. Electrolysis of the molten oxide is required.

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New 21st Century Chemistry b) c) d) Discussion (page 177) This discussion aims at raising students ' awareness of the principles of green chemistry. See the following list of green chemistry principles the discussion questions help to convey. 1 Safer auxiliary substances — Whenever possible, avoid the use of auxiliary substances such as

~~Suggested answers to in-text activities and exercises~~

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New 21st Century Chemistry Suggested answers to in-text activities and unit-end exercises Topic 10 Unit 36 In-text activities Checkpoint (page 6) 1 Number of moles of H2 formed = = 0.00800 mol Average rate of formation of H2 = = 2.00 x 10 – 4 mol s – 1 2

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