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PART-1 6.1 Before solving this
question, examine ex.6.1 in
Lamarsh and for cross sections
and other constants use table 6.1;

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0.03 32.33 Pu Na Pu Na Pu and
then; ,,,,, 0.03 32.33 1 1 0.108
and 2.61 * 1 1 * 2.11 8 4 and
finally k 2.222(sup) Pu Na Pu Na
Pu Na a Na Na a Na Na Pu a Pu Pu
a Pu Pu a Pu a Na f N Navgdro M
N ...

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PART-1 6.1 Before solving this question, examine ex.6.1 in Lamarsh and for cross sections and other constants use table 6.1; 0.03 32.33 Pu Na Pu Na Pu and then; , , , , 0.03 32.33 1 1 0.108 and $2.61 * 1 1 * 2.11 8 4$ and finally $k 2.222(\text{sup})$ Pu Na Pu Na

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Solutions Chapter-2 2.5 This is a
question of probability, For

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molecules which have an approximate weight of 2, there are two ^1H and we can find the probability or the percentage over 1 as, $0.99985 \times 0.99985 = 0.99970$
The same calculation can be made for the mol.

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that distractor valence would interact with prior and current flanker congruency, but showed a trend toward an effect of distractor valence on conflict adaptation.

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SOLUTIONS CHAPTER-7 PART-1

7.1 Look at example 7.1 in the
textbook, only the moderator
materials are different Since the
reactor is critical, $k_{Tf} = 1$ $T = 2.065$

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from table 6.3 so $f = 0.484$ We will use $t_{d,TM}(1-f)$ and $t_{d,TM}$ from table 7.1 $t_{d,TM,D2O} = 4.3e-2$; $t_{d,TM,Be} = 3.9e-3$; $t_{d,TM,C} = 0.017$ Then, $t_{d,D2O} = 0.022188\text{sec}$; $t_{d,Be} = 2.0124e-3\text{sec}$; $t_{d,C} = 8.772e \dots$

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