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Answers for the questions in Chapter 8: ROUND 3

Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions.

Use abbreviations to indicate the appropriate phase of reactants and products where possible. Note: Not all of the reactions will occur. For those that do not, write no reaction. I.

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Answers for the questions in Chapter 8: ROUND 3

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Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions. Use abbreviations to indicate the appropriate phase of reactants and products where possible.

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Determine the oxidation number of the underlined element: KMnO_4 Since K is an alkali metal, its charge must be $1+$ Oxygen is $2-$ but there are four of them, therefore, 4 times $2-$ equals $8-$ If $1+$ and $8-$ are added Ultimate Chemical Equation Answers

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The Ultimate Chemical Equations Handbook Answers Chapter 7

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Quiz on all Chemical Reactions. Pages are from the Ultimate Chemical Equations Handbook. •

Precipitation Reactions(See solubility rules) - Pg. 28, Practice-Pg. 49 • Formation of a Gas - Pg. 50, Practice-Pg. 51 • Acid Base Neutralization Reactions- Pg. 53, Practice-Pg.54 oWeak acids and bases do not break apart and are in net ionic equation. • Synthesis and Decomposition Reactions - Pgs. 42-43, Practice- Pg. 43 • Single Replacement Reactions - Pg. 45, Practice-Pg. 46.

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